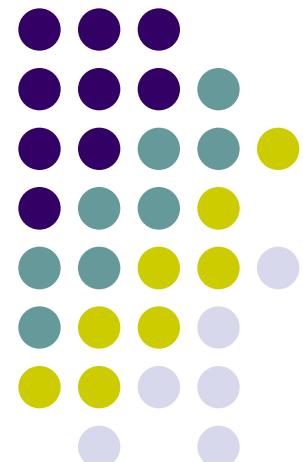




Tetraphenylporphyrin/Polyaniline Complexes as Optic Active Layer in Organic Optoelectronic Applications

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³Kelompok Keahlian Fisika Nuklir dan Biofisika, Jurusan Fisika, Fakultas Matematika dan Ilmu Pengetahuan Alam, Institut Teknologi Bandung, Jalan Ganesha 10, Bandung 40132, Indonesia

⁴Kelompok Keahlian Teknik Fisika, Jurusan Teknik Fisika, Fakultas Teknologi Industri, Institut Teknologi Bandung, Jalan Ganesha 10, Bandung 40132, Indonesia

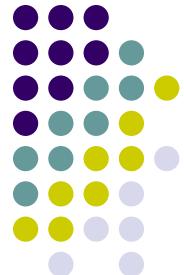
⁵National Research Centre for Nanotechnology, Jalan Ganesha 10, Bandung 40132, Indonesia

⁶Institute of Industrial Science, The University of Tokyo, 4-6-1 komaba meguro-ku, tokyo 153-8505, Japan



Plan of Presentation

- Introduction
- Experiments
- Results and Discussion
- Conclusions

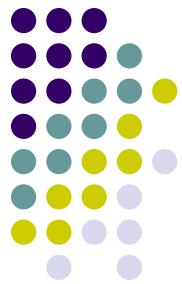


Tetraphenylporphyrin/Polyaniline Complexes as Optic Active Layer
in Organic Optoelectronic Applications

INTRODUCTION

Global Energy Demand





Solar Energy



Obstacles :

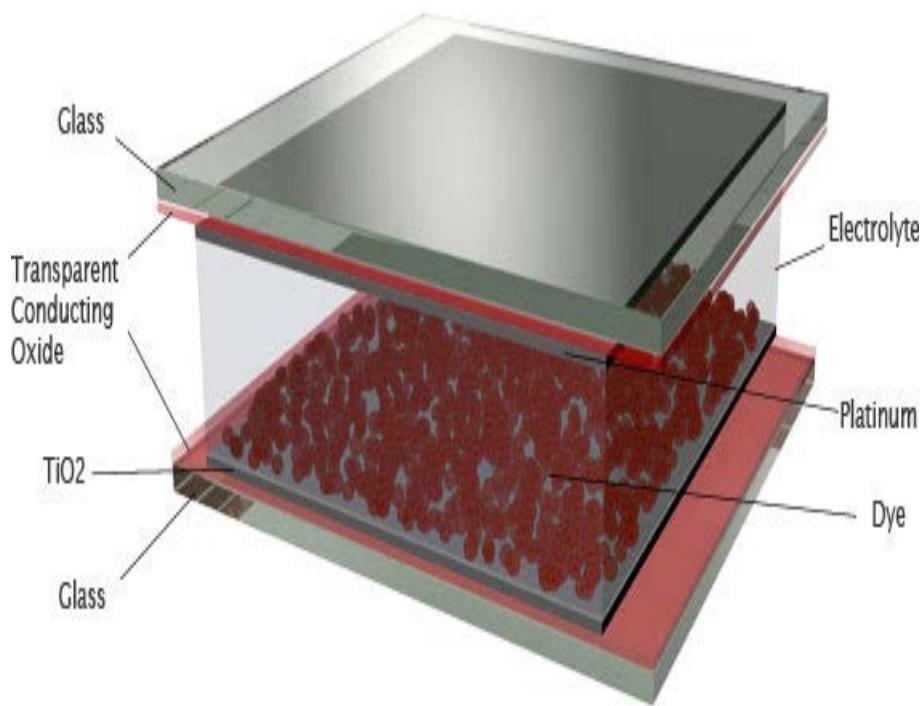
- High production cost
- Expensive components
- All components are imported

A cheaper photovoltaic technology has to be developed in Indonesia



DSSC (Dye-Sensitized Solar Cell)

- Simple Fabrication
- Dye/sensitizer can be obtain locally: natural pigments, i.e. photosynthetic pigments from green plants, anthocyanins, etc.
- Very suitable to be implemented in Indonesia.

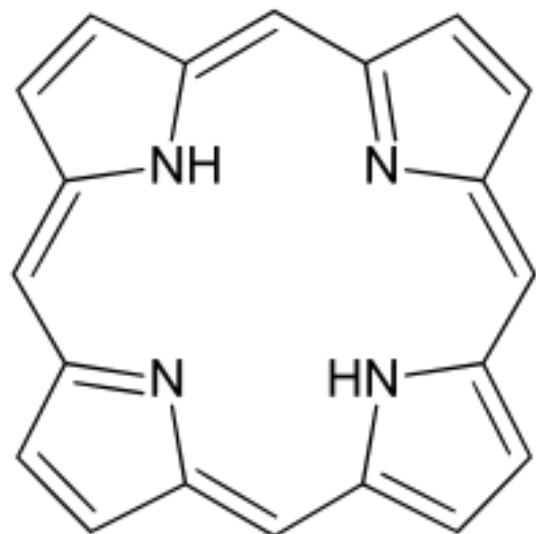


DSSC components



Porphyrin

- A potential dye/sensitizer for DSSC that covers a certain range of light radiation
- High quantum efficiency
- Has a unique molecular junction with a wide-band gap semiconductor that depends on how it was immobilized
- High molecular symmetry with point group of D_{4h} (metalloporphyrin) or D_{2h}/C_{2v} (free-base porphyrin)

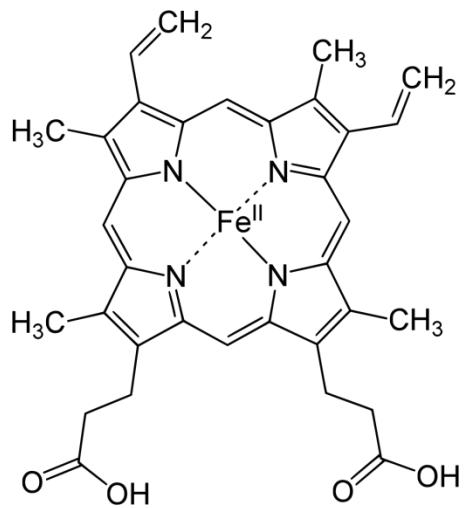
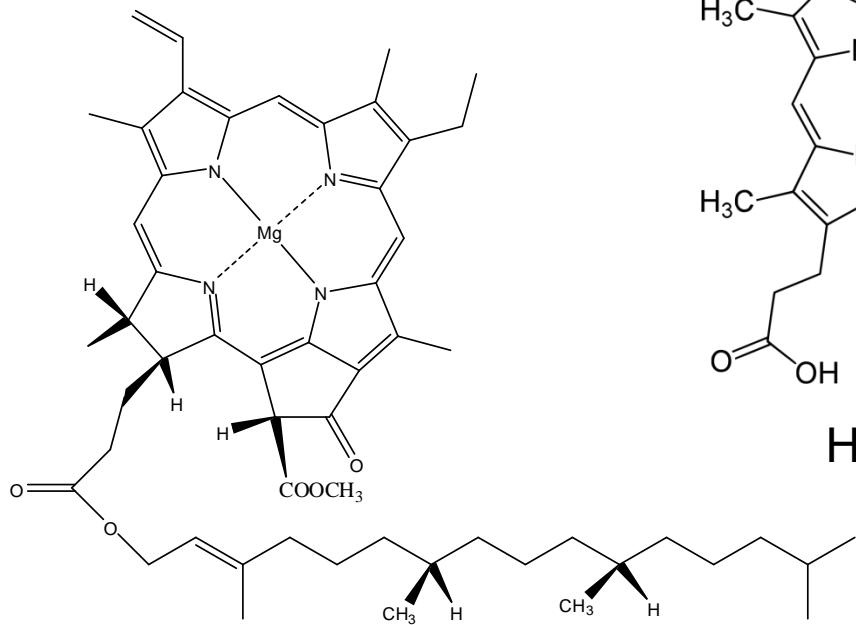




Porphyrin

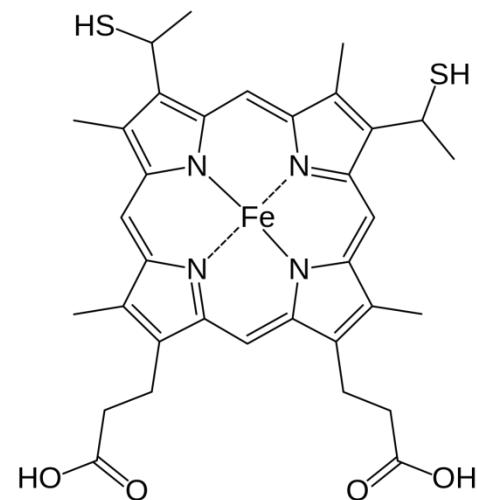
- Several porphyrins are available abundantly in nature: chlorophyll a, heme B (in hemoglobin), heme C (pada *cytochrome c*).

chlorophyll a

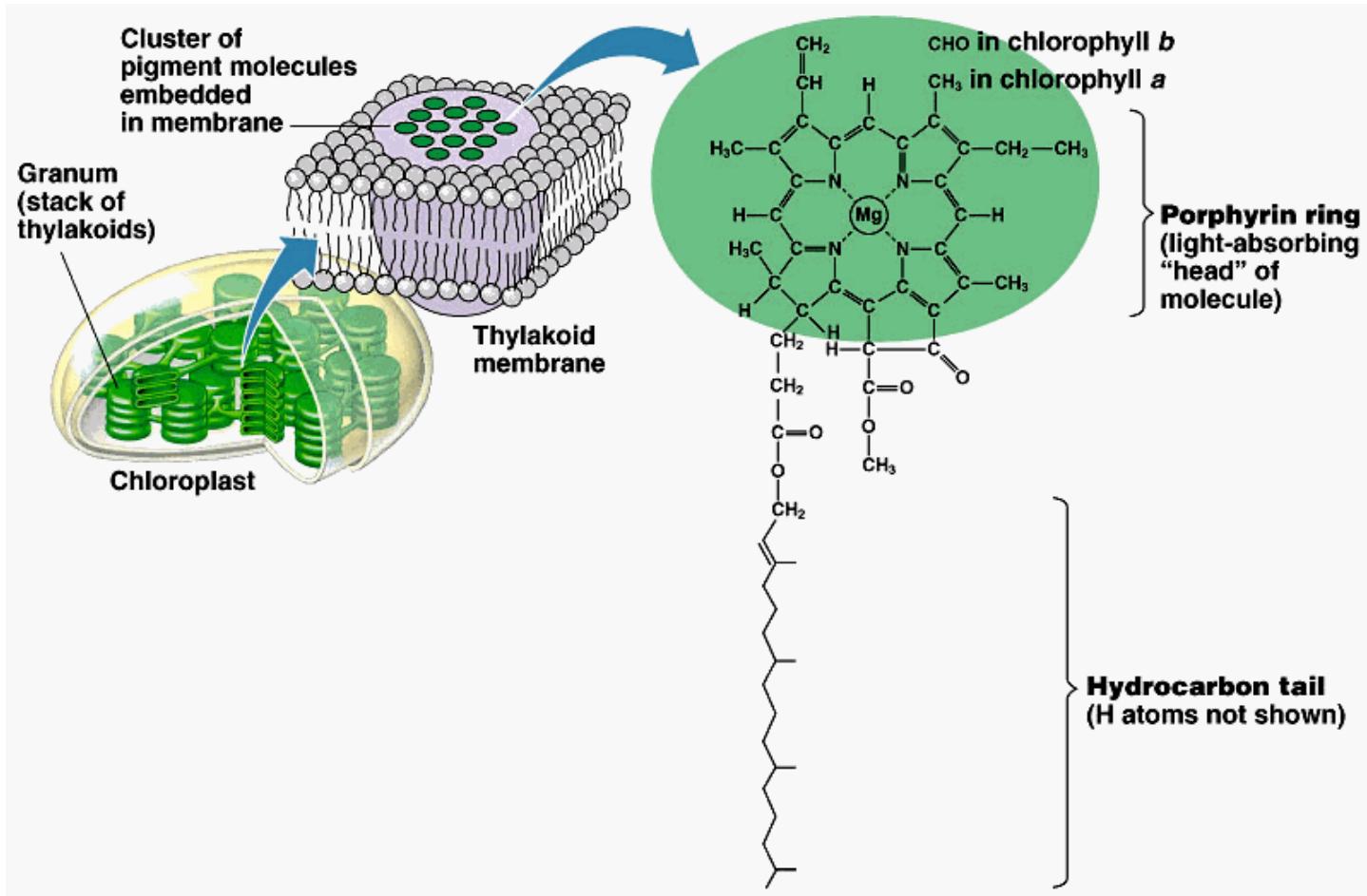


Heme B

Heme C



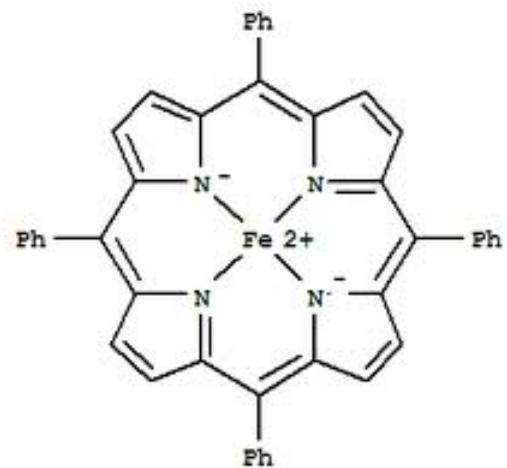
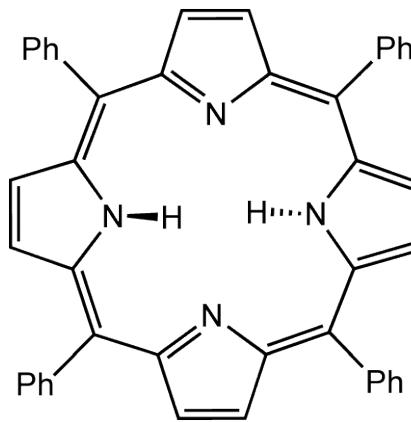
Chlorophyll

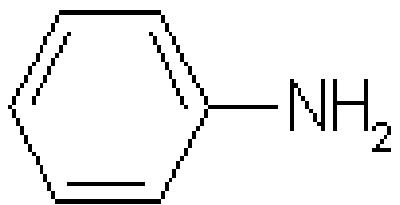




Meso-Tetraphenylporphyrin (TPP)

- Synthetic pigment with a simple synthesis pathway and high yield
- High absorption coefficient in visible light
- Gives a strong photoluminescence in red
- Simple to vary its properties through the insertion of metal ions, protonation and addition of linker molecules





Aniline

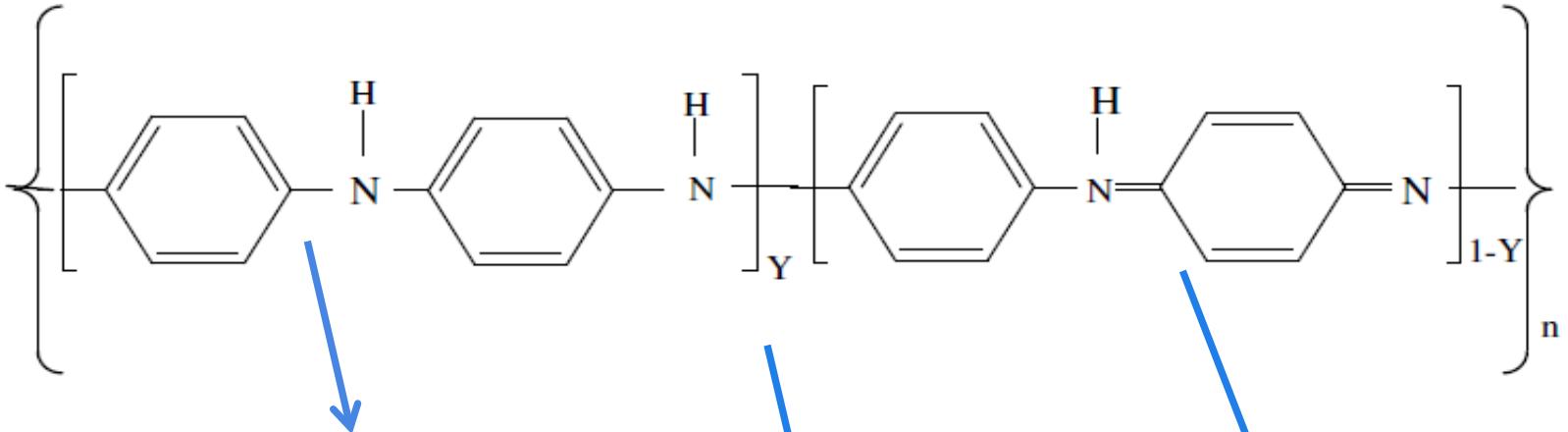
Polyaniline

Conductive polymer

Advantages :

- Low production cost
- Covers a wide range of electrical conductivity (10^{-10} S/cm - 100 S/cm)
- High environmental and thermal stability

POLYANILINE

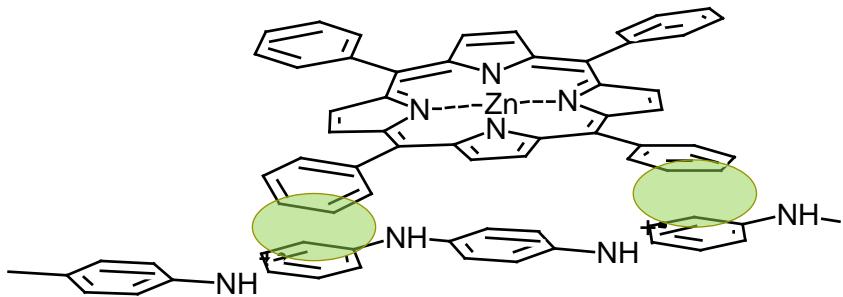


Leucoemeraldine
Base (LEB), $Y = 1$

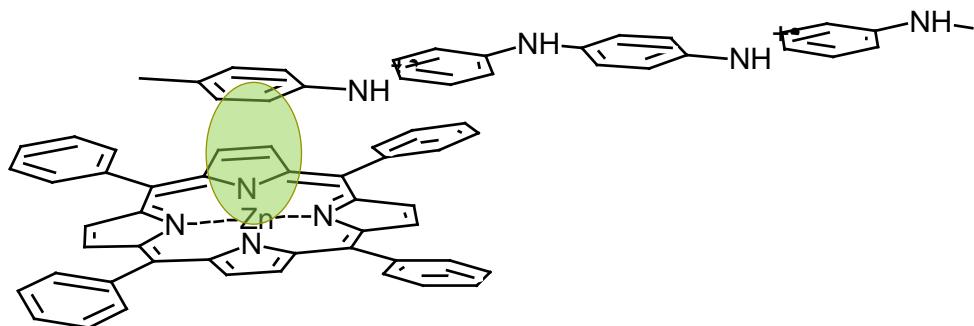
Emeraldine Base
(EB), $Y = 0.5$

Pernigraniline
Base (PNB), $Y = 0$

Molecular Junction in TPP/PANI Blend

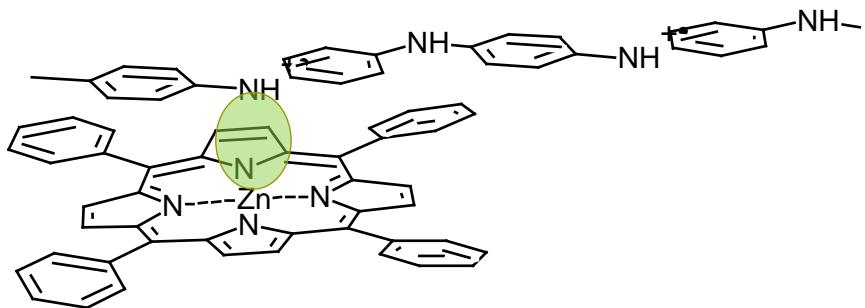
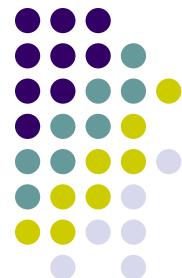


π Stacking interaction

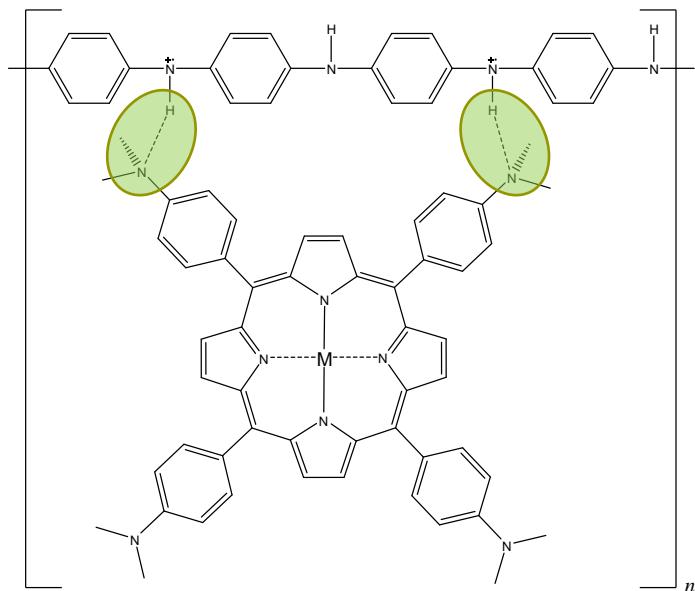


**covalent coordination
with type (η^6 , L₃), (η⁴,
L₂) or (η², L), between
π electrons and
central metal ion**

Molecular Junction in TPP/PANI Blend



covalent coordination
with type (η^2 , L)
between *lone pair* and
central metal ion



Lewis acid-base interaction

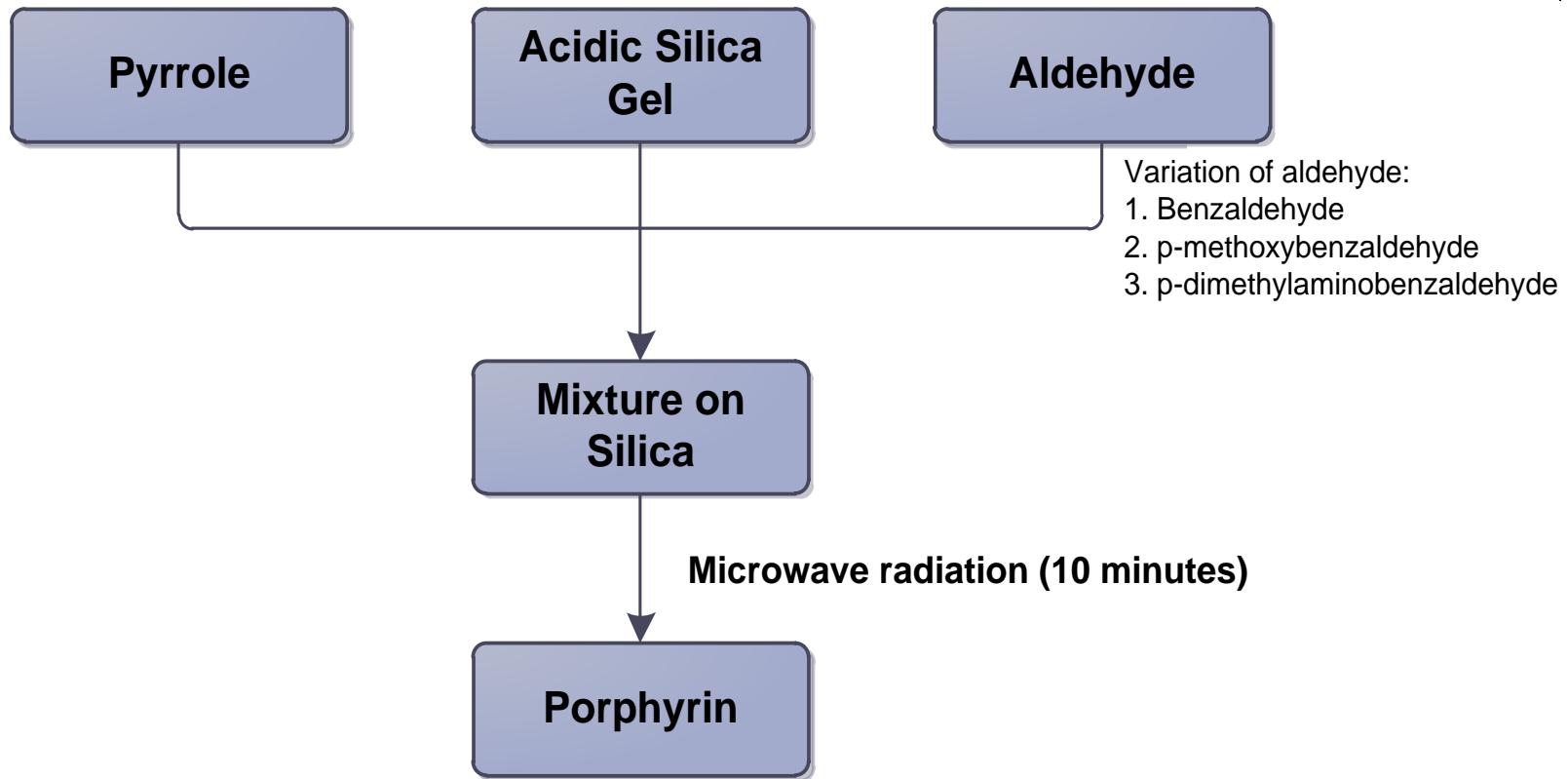


Tetraphenylporphyrin/Polyaniline Complexes as Optic Active Layer
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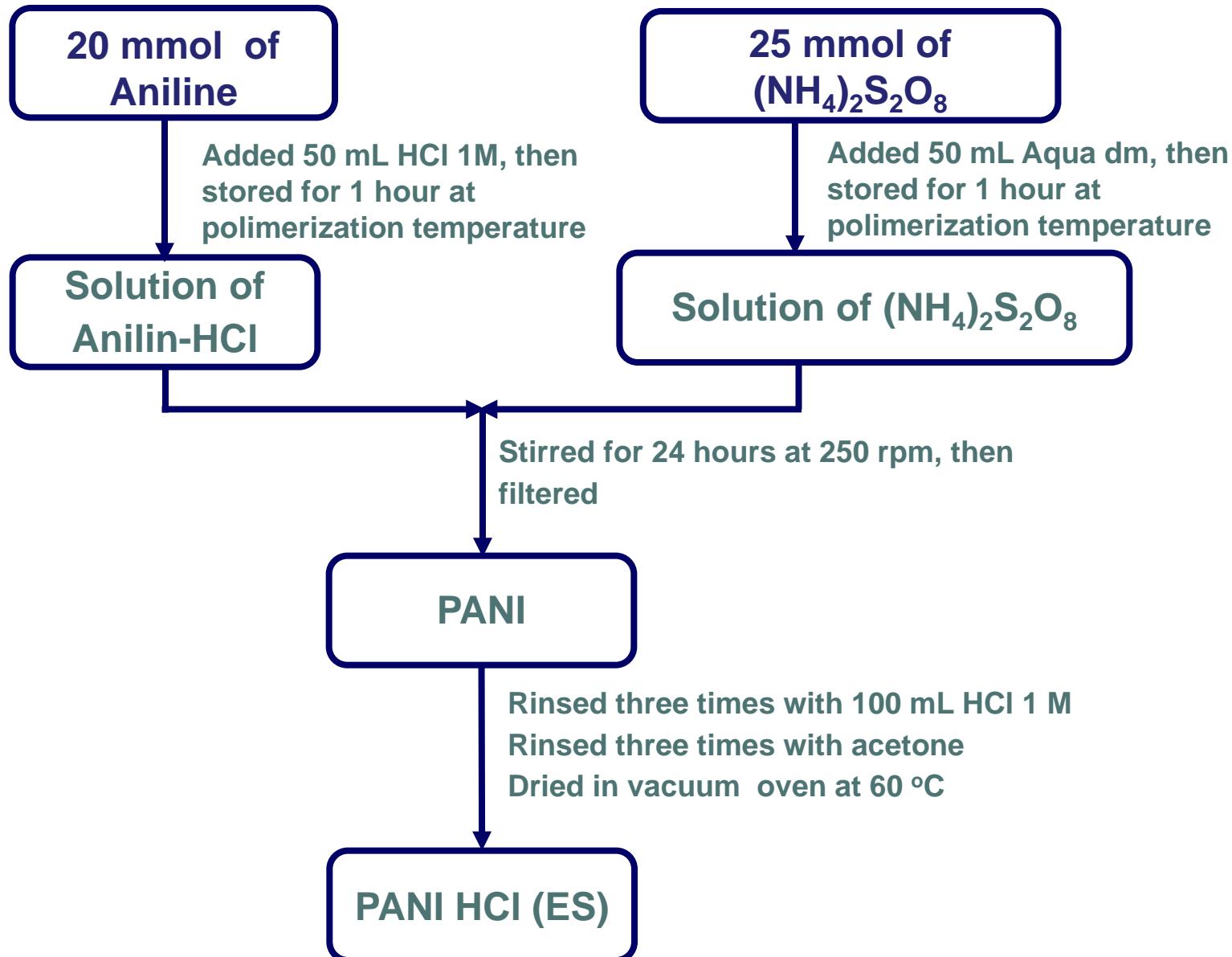
EXPERIMENTS



Synthesis of Porphyrins

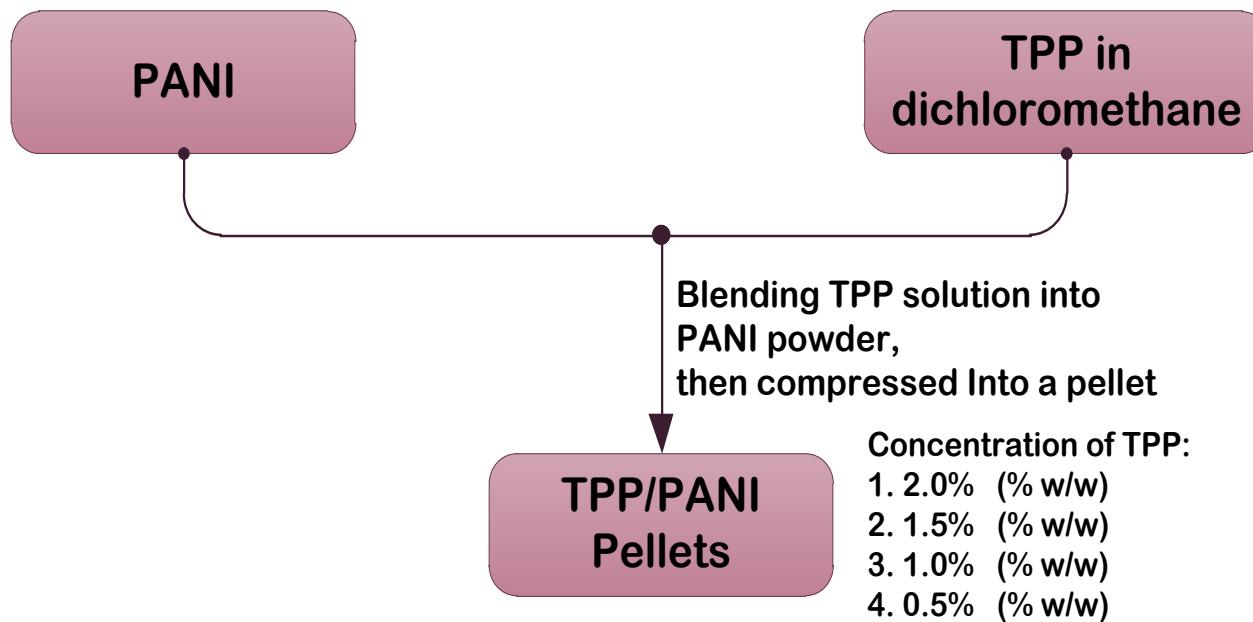


Synthesis of Polyaniline



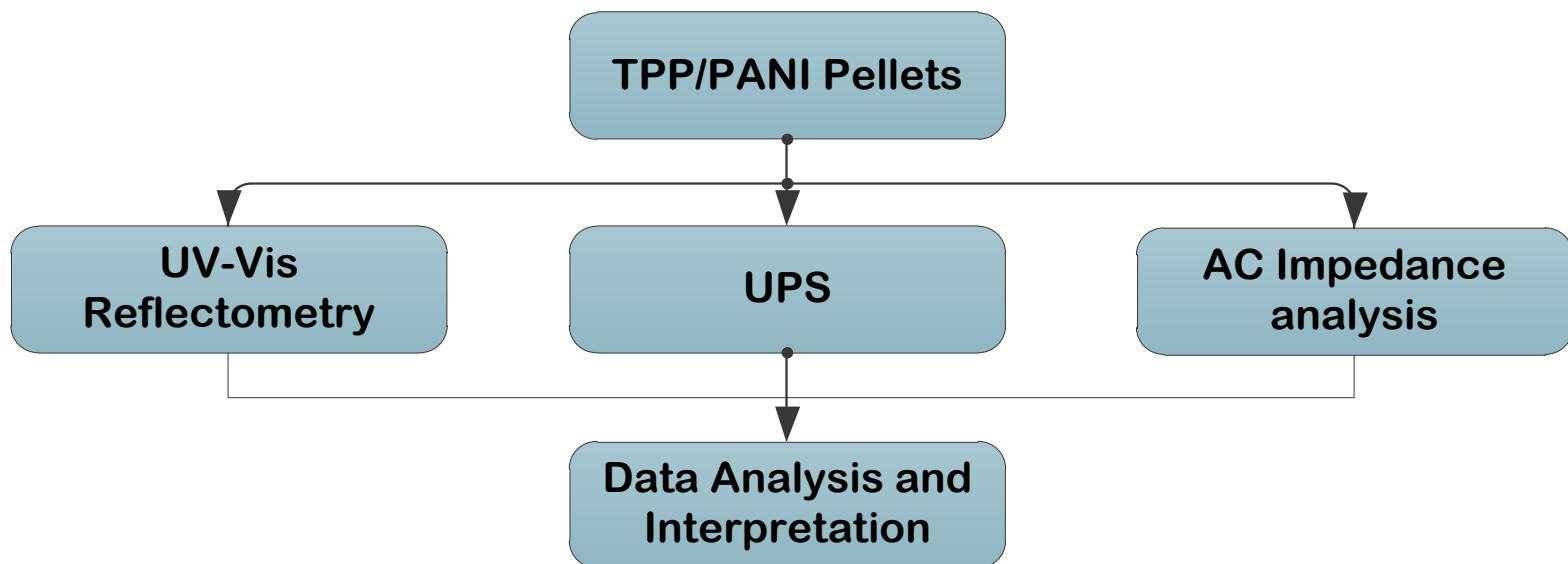


Blending of TPP/PANI

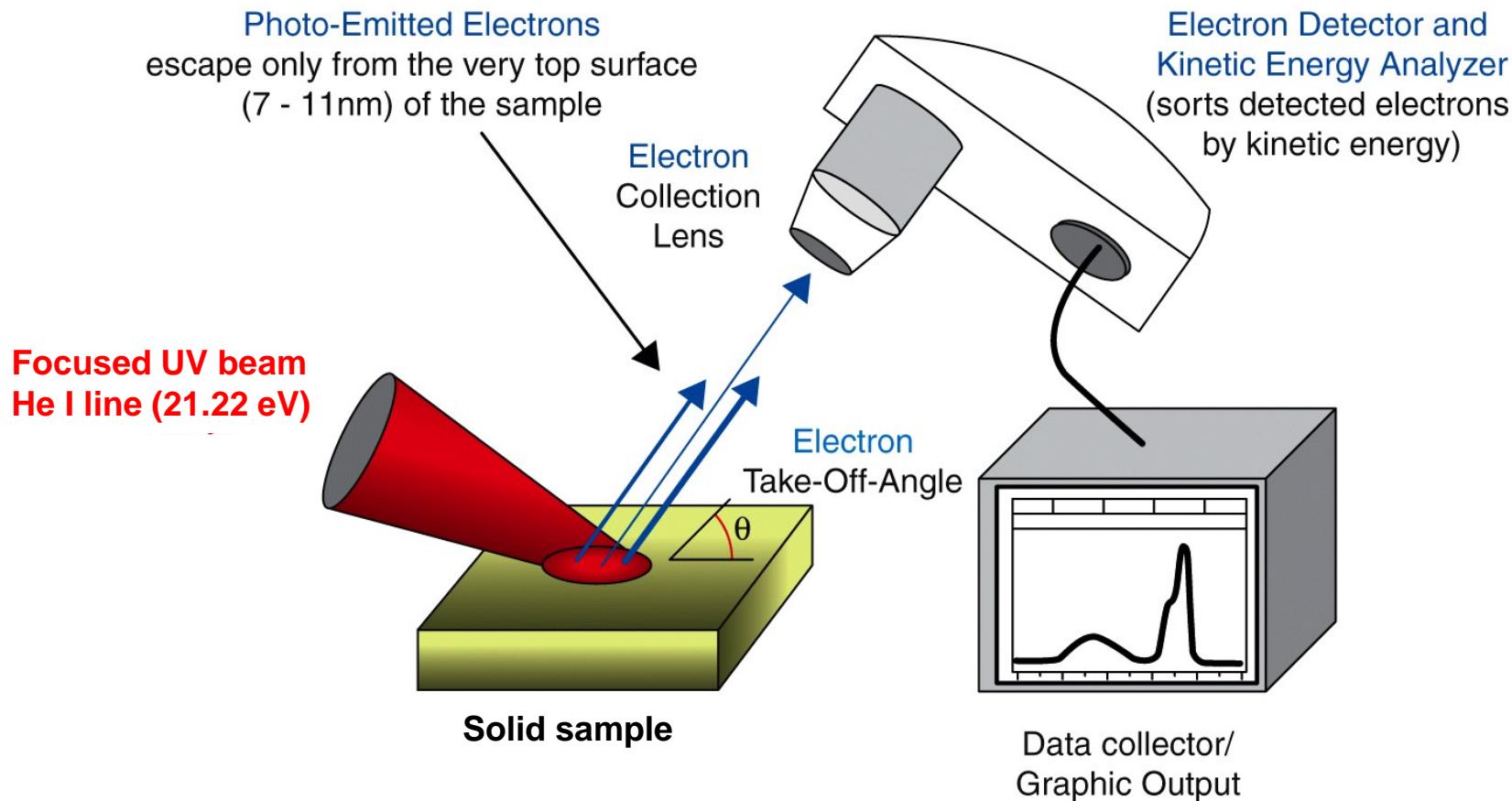




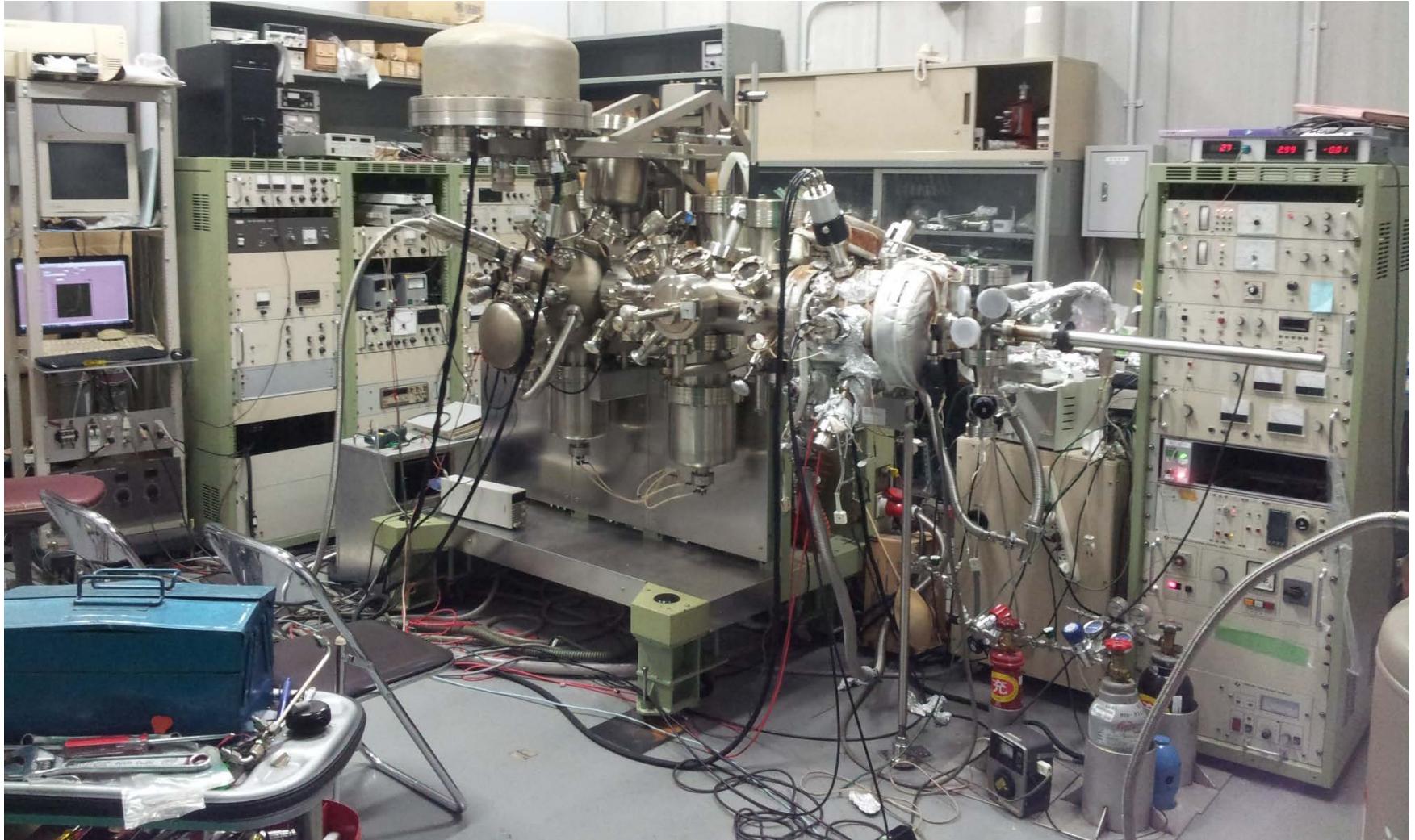
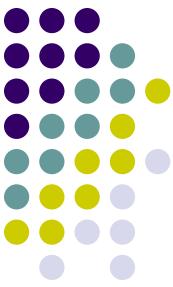
Characterization of TPP/PANI



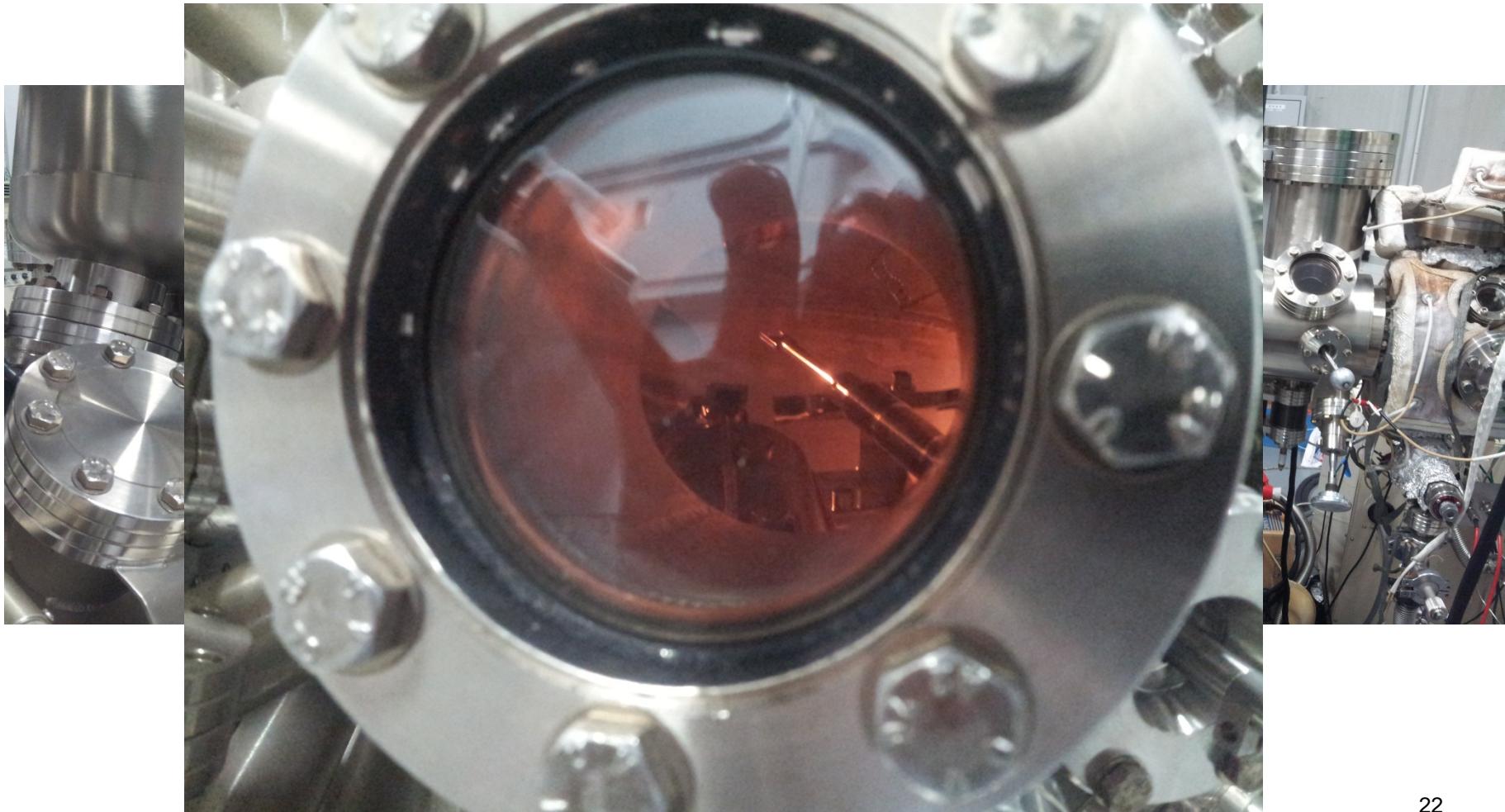
Schematic of UV Photoelectron Spectroscopy (UPS)



UV Photoemission Spectrometer in IIS/The University of Tokyo



UV Photoemission Spectrometer in IIS/The University of Tokyo



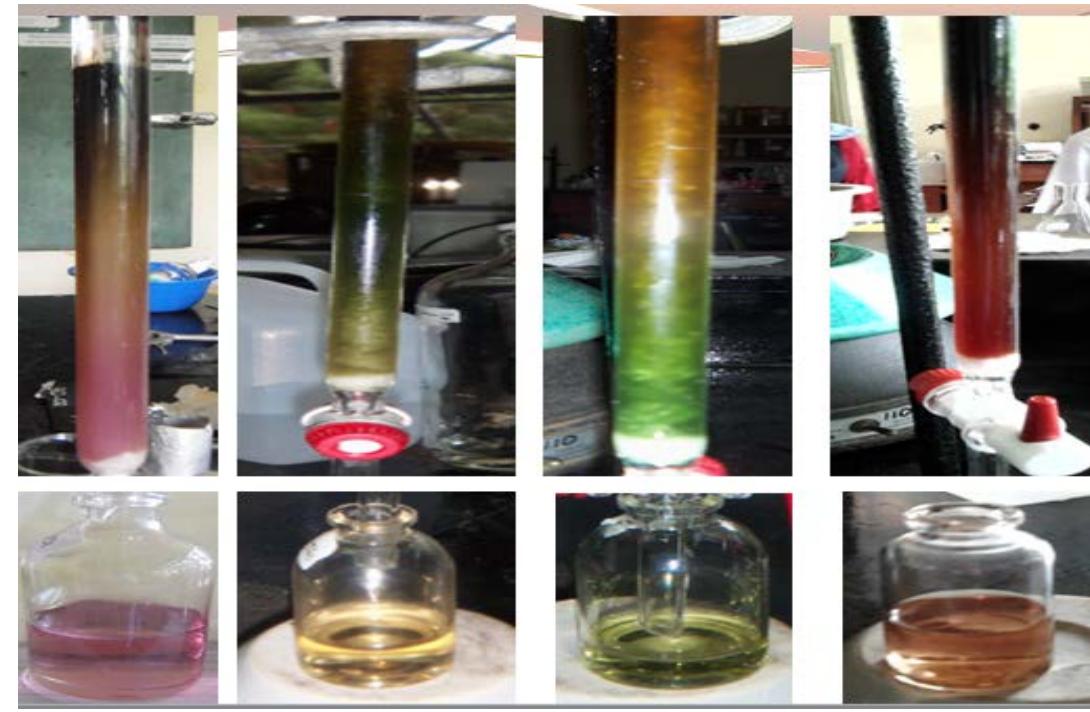
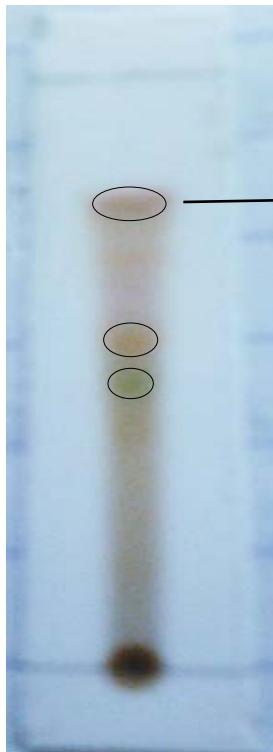


Tetraphenylporphyrin/Polyaniline Complexes as Optic Active Layer
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RESULTS AND DISCUSSION

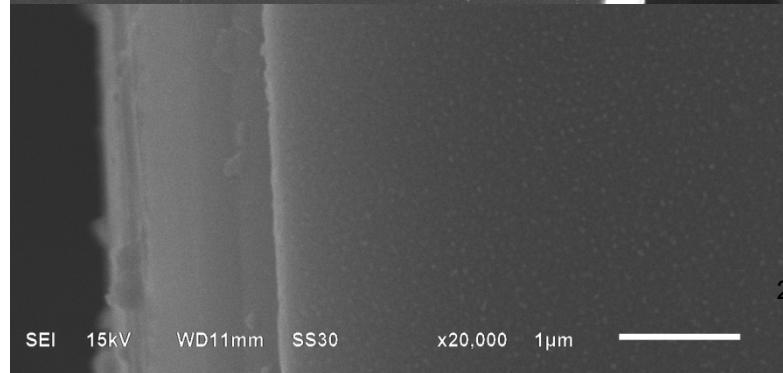
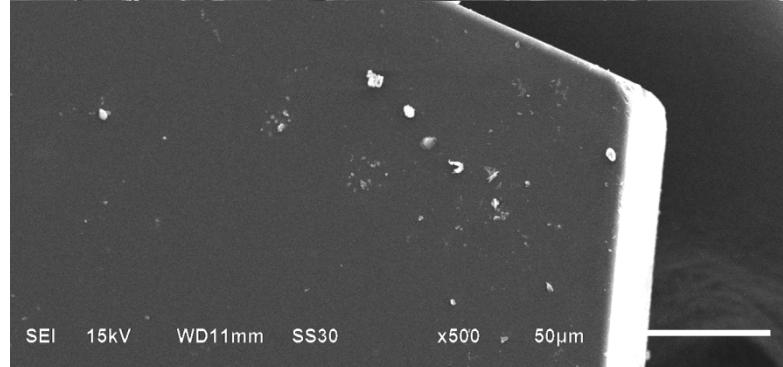
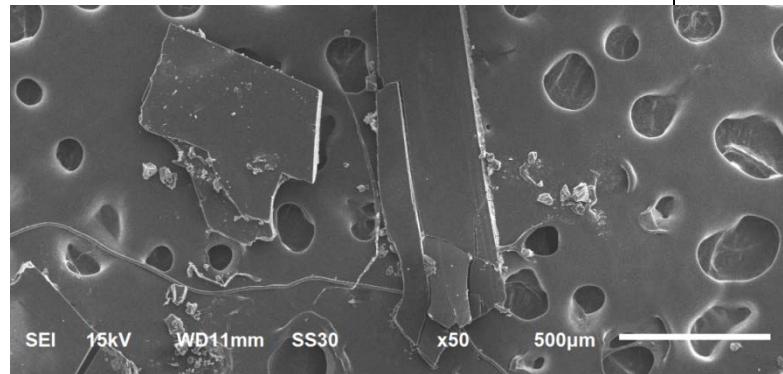
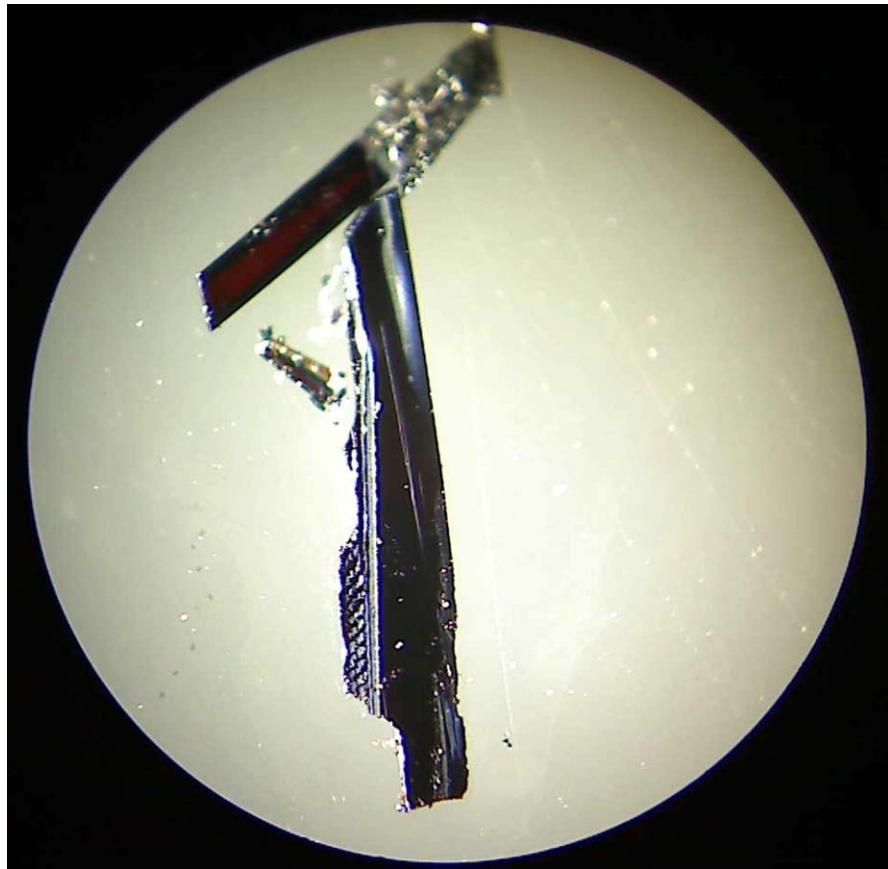


meso-Tetraphenylporphyrin synthesis

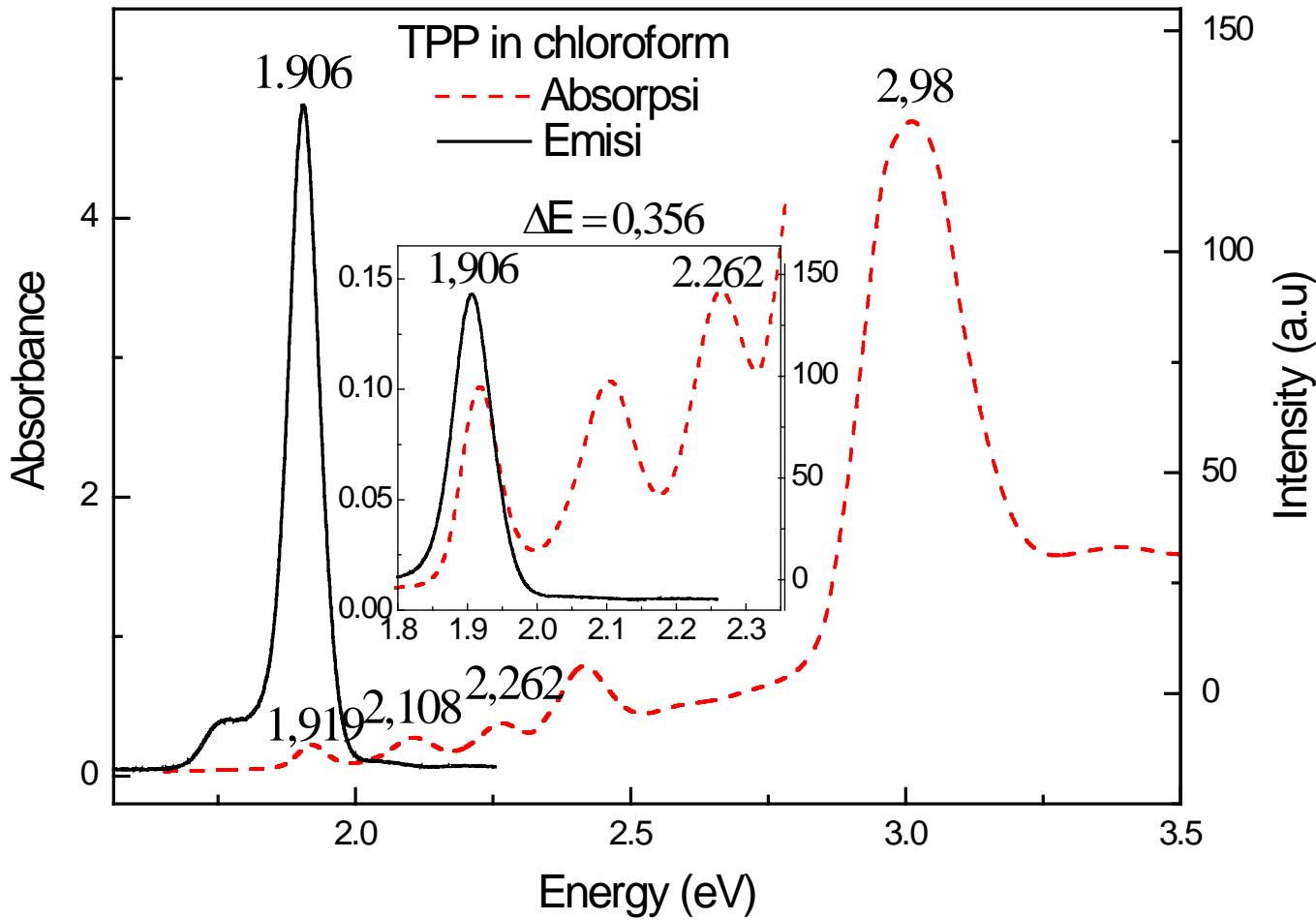




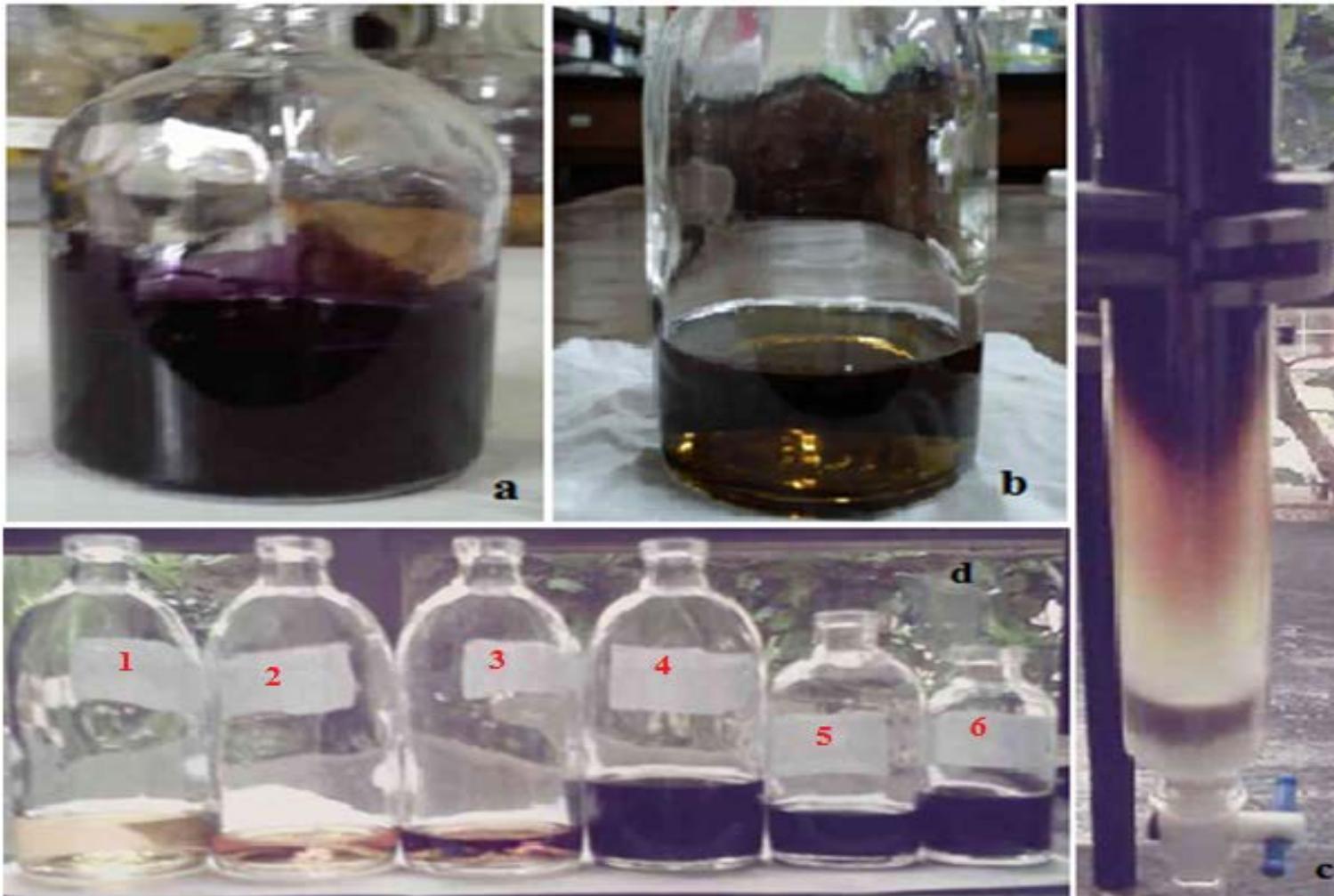
meso-tetraphenylporphyrin crystals



UV-VIS absorption and emission spectra of TPP in chloroform



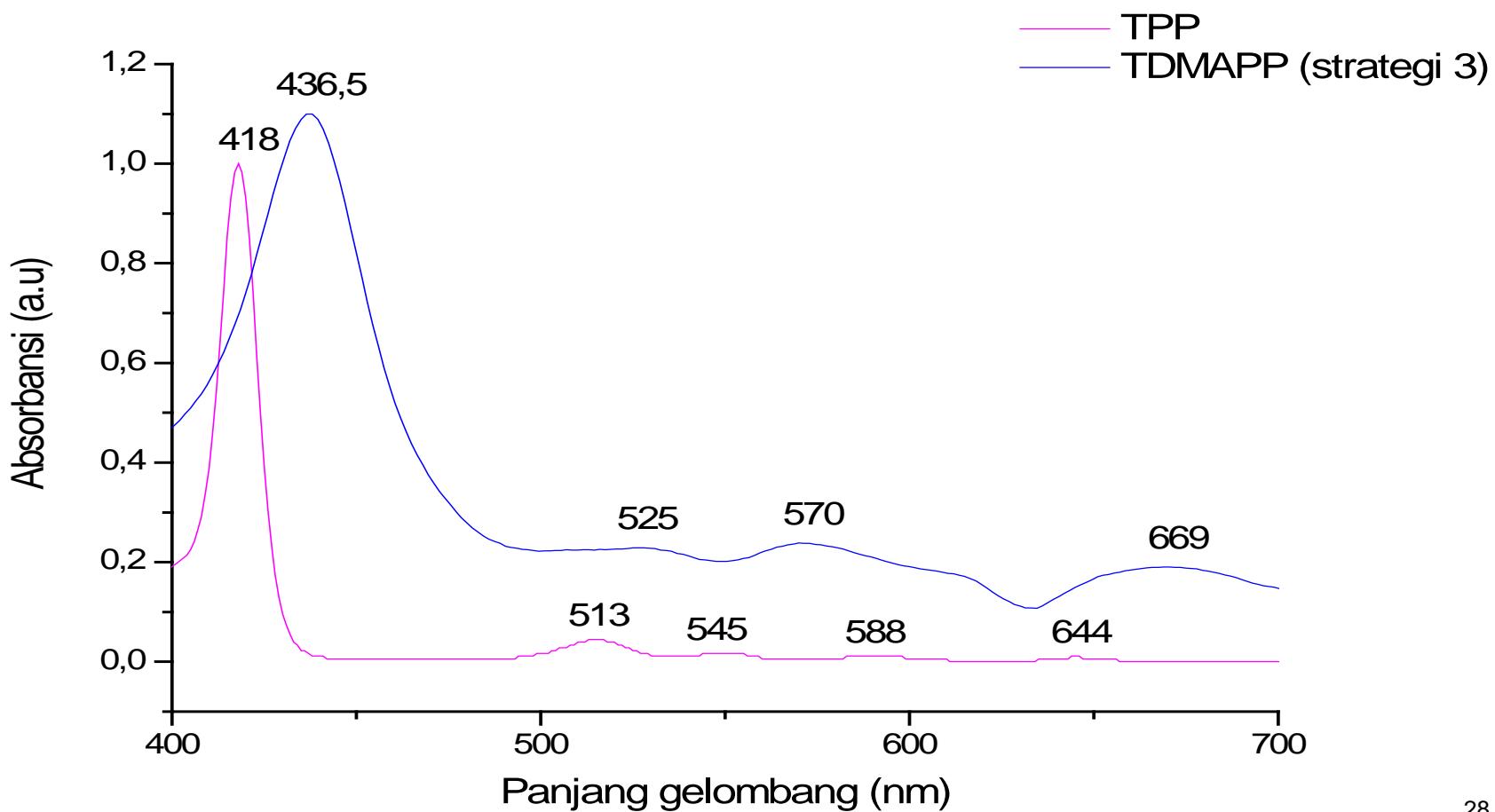
*meso-Tetra(*p*-dimethylaminophenyl)porphyrin synthesis*



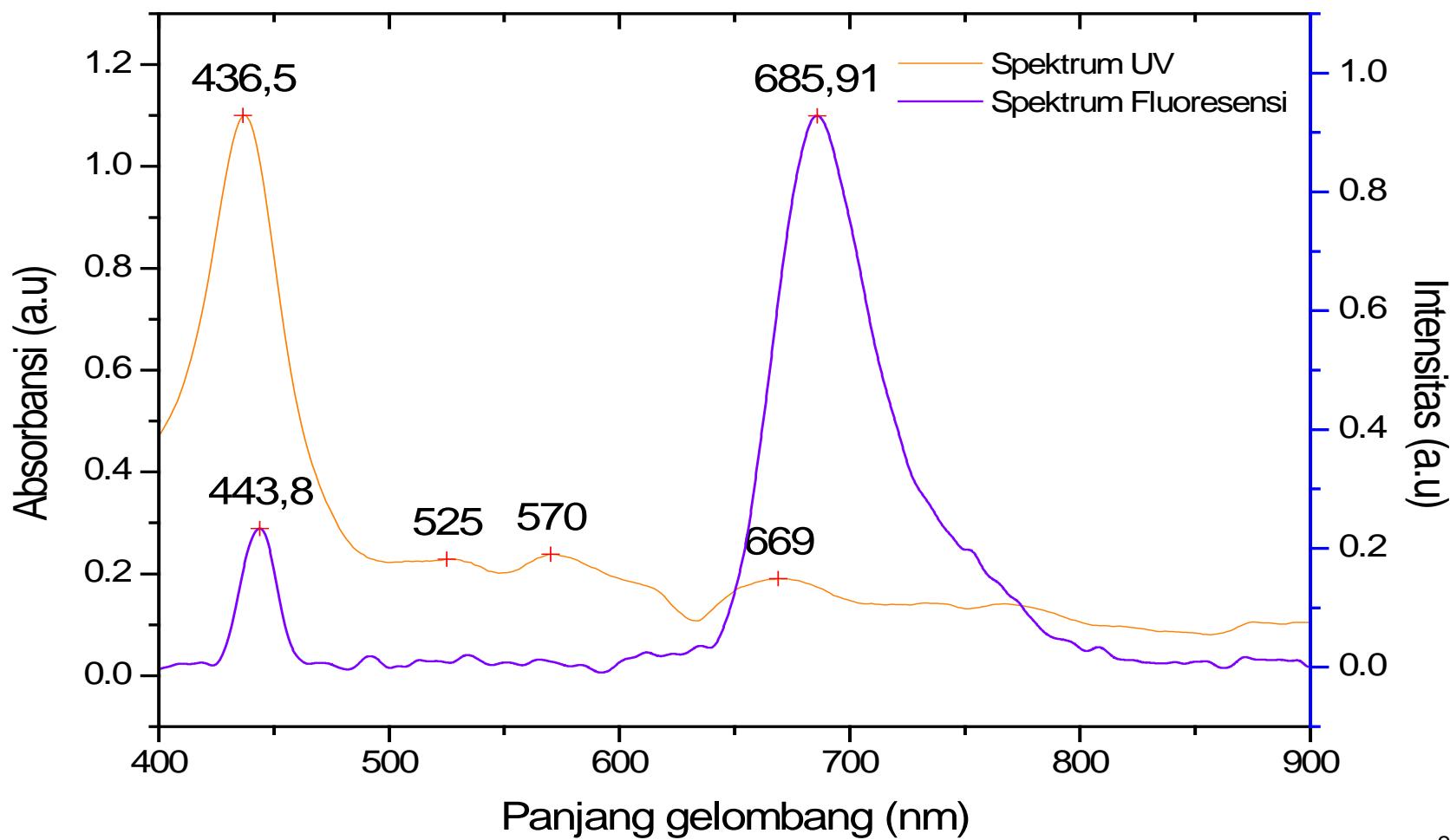
a) As synthesized in silica gel , b) crude extract in chloroform, c) purification using column chromatography, d) several fractions after purification.



meso-Tetra(*p*-dimethylaminophenyl)porphyrin synthesis

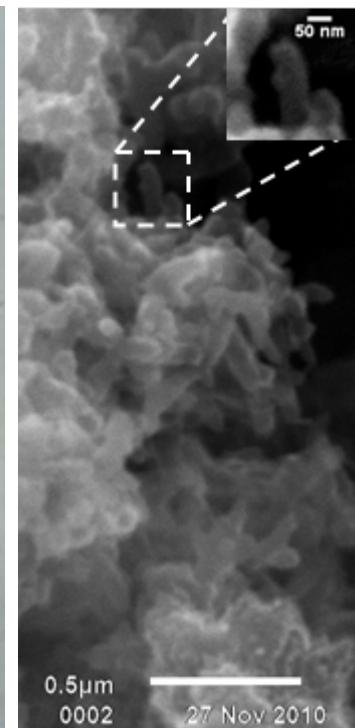
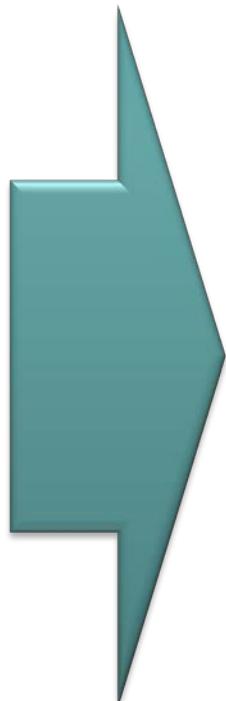


meso-Tetra(*p*-dimethylaminophenyl)porphyrin synthesis



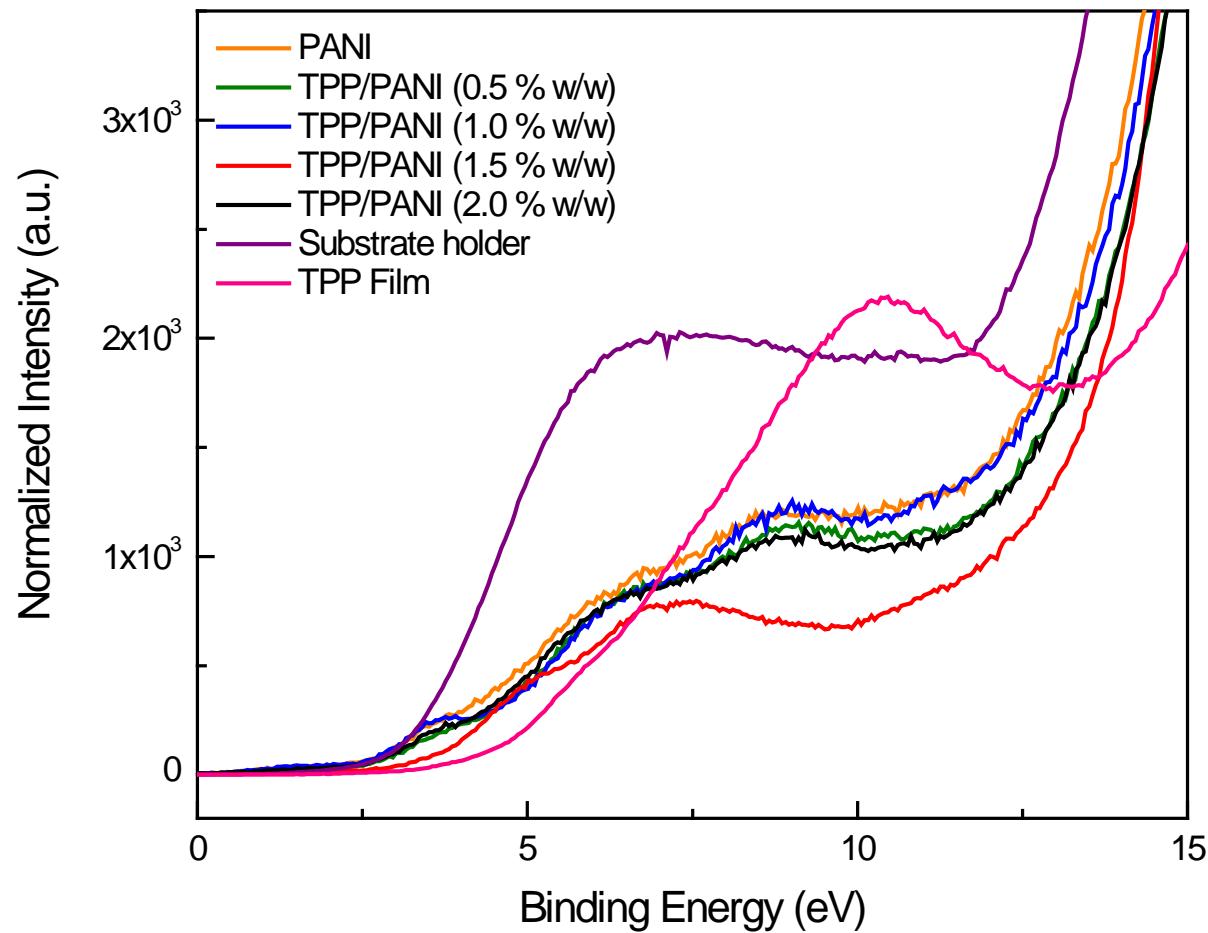


Polyaniline synthesis



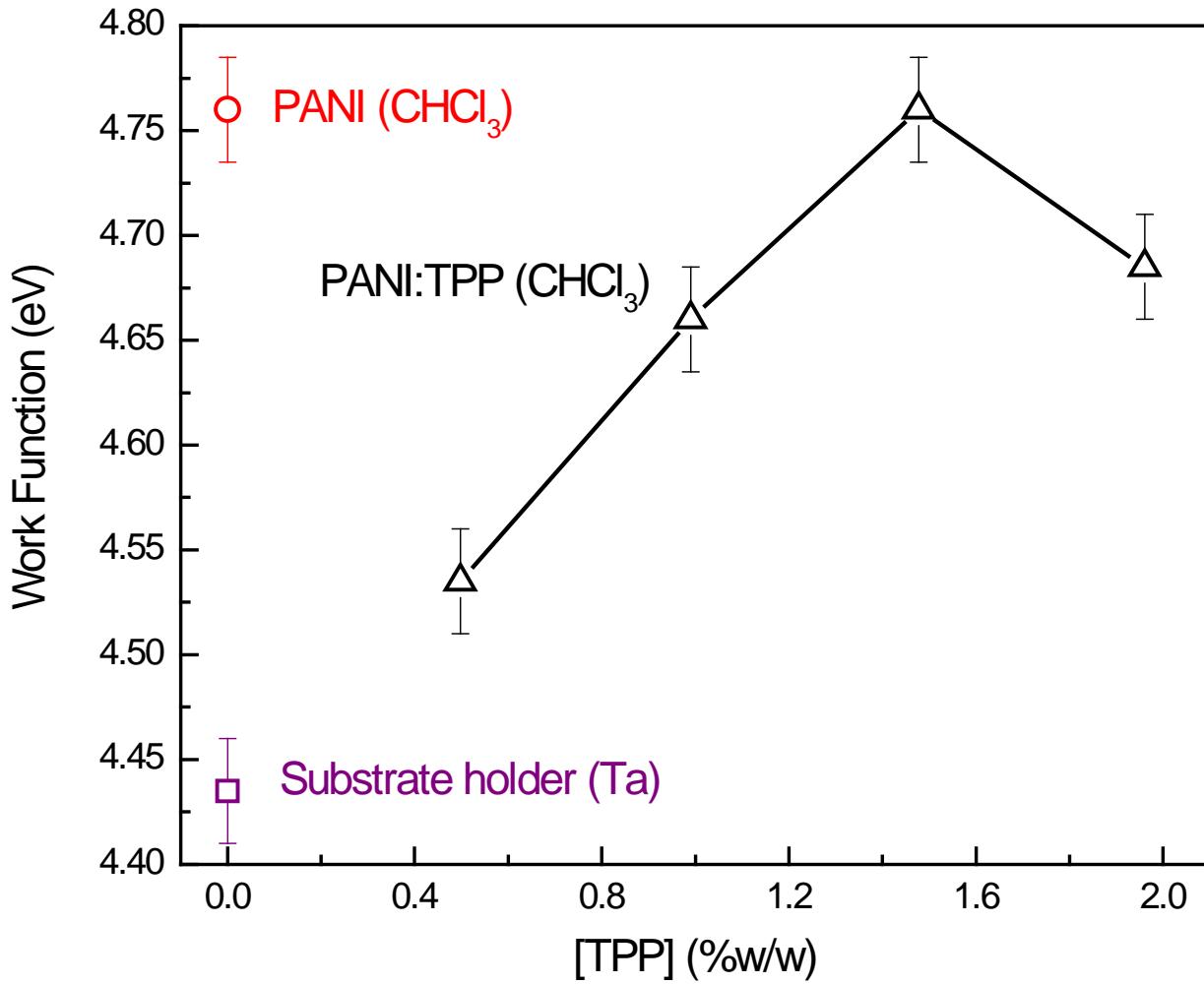


UV Photoelectron spectra



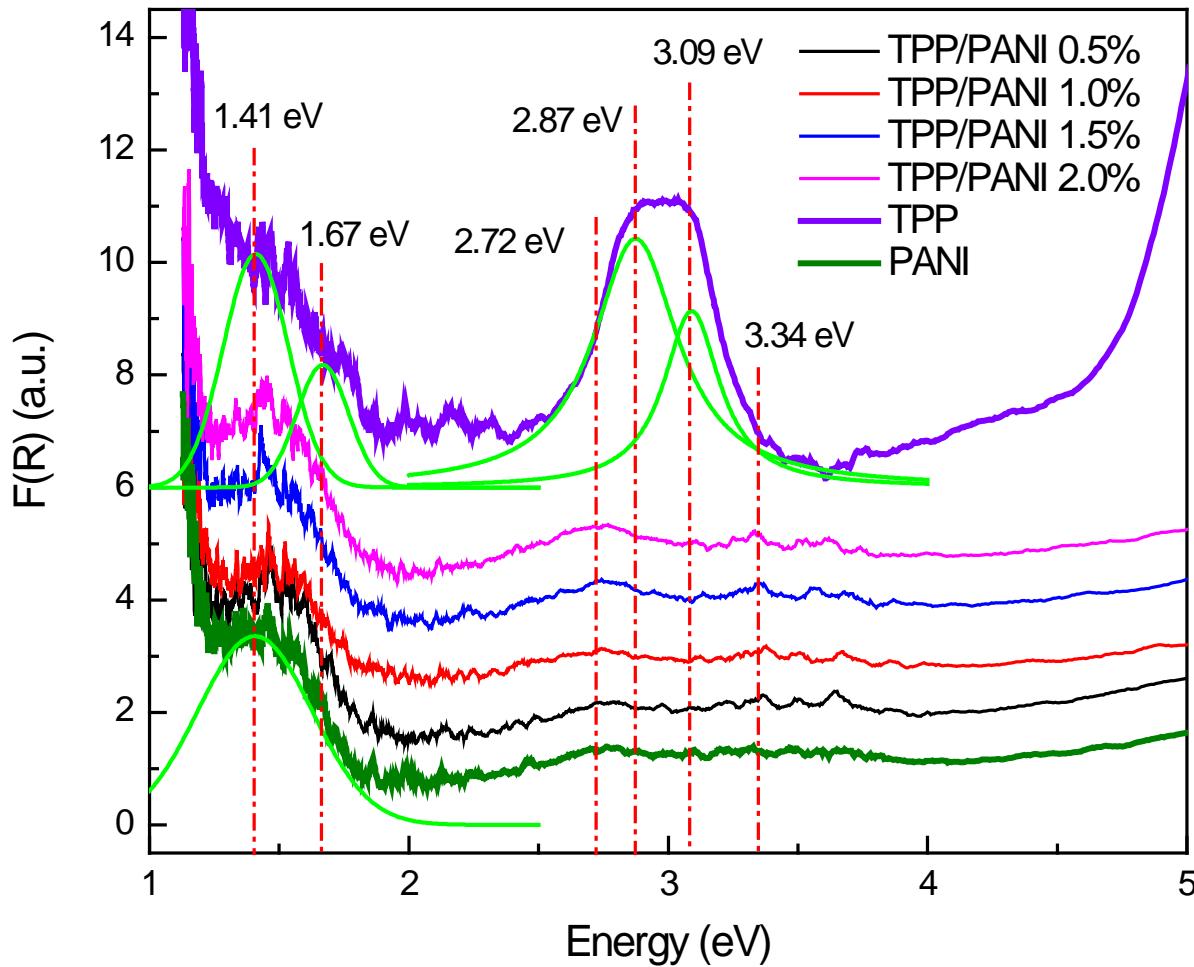


Surface workfunction



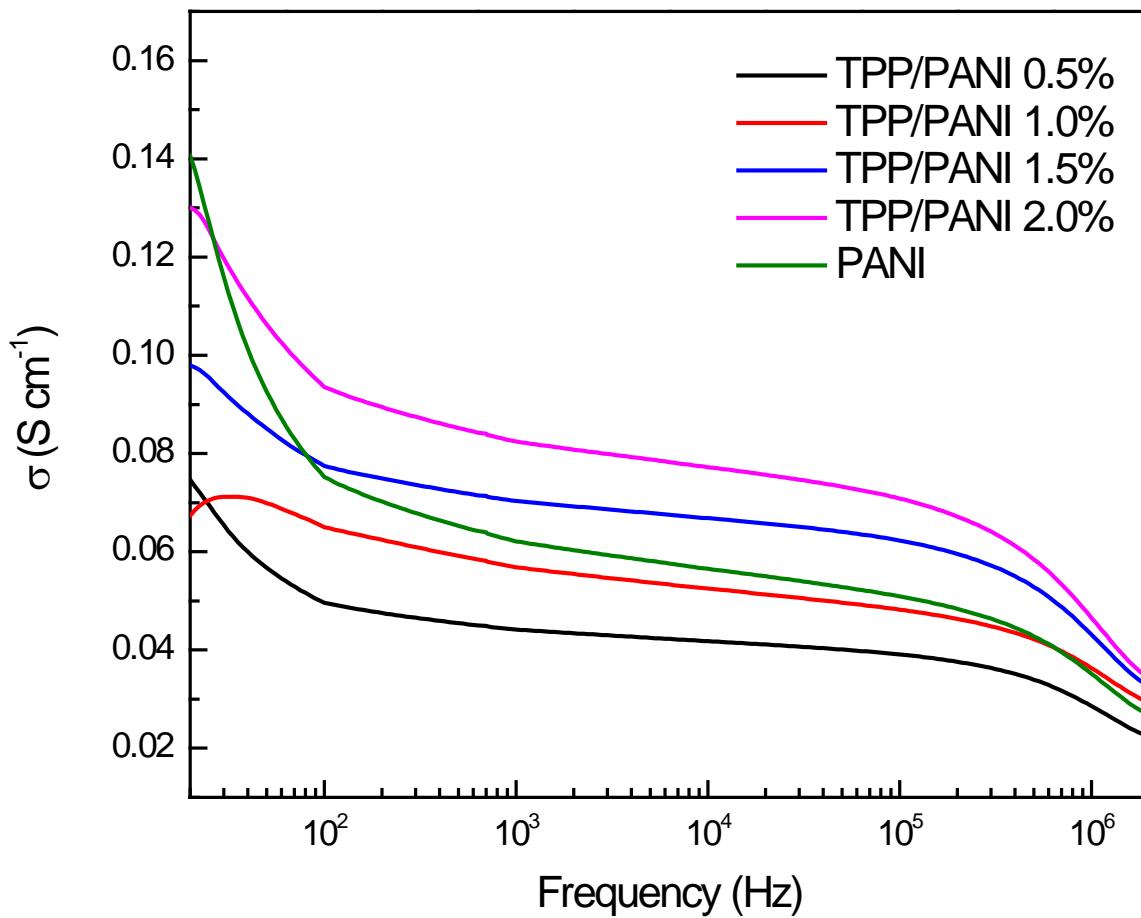


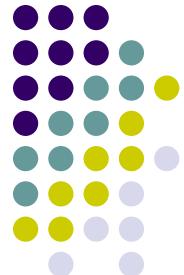
UV-Vis reflectance spectra





Electrical conductivity





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CONCLUSIONS



Conclusions

- Both UPS and UV-Vis reflectance results show that the presence of TPP in TPP/PANI blend adds some electronic states in PANI valence band.
- A certain concentration limit is necessary to alter the properties of the blend significantly with respect to the pure PANI.
- Interesting doping feature was also observed in the electrical conductivity as a function of frequency of TPP/PANI blend, where the presence of TPP improved the high frequency part of electrical conductivity significantly.



Future Plans

- Ab initio study on the interaction between TPP and PANI in TPP/PANI blend and its electronic properties.
- Photoluminescence measurements of TPP/PANI blends in their resonance wavelengths.
- Raman/FTIR measurements to reveal the interaction between TPP and PANI in TPP/PANI blend
- Applying TPP/PANI blends as a simple PV device.



Thank you

